

## PCP5-4

### Synthesis of the Mother Phase of the Iron-Based Superconductor, SmFeAsO via Low-Temperature Heat Treatment

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Low-temperature heat treatment of the iron-based superconductor<sup>1,2</sup> is effective for fabrication of the iron-based superconducting wires and tapes with high transport critical current density. To determine the lowest temperature for the formation of a crystallographic phase of SmFeAsO, we demonstrate the evolution of the SmFeAsO during a solid-state reaction in a mixture of SmAs, Fe<sub>2</sub>As, FeAs, and Sm<sub>2</sub>O<sub>3</sub> heated to heat-treatment temperatures from 580°C to 950°C. X-ray diffraction (XRD) measurements indicated that a significant increasing of the SmFeAsO phase appears at heat-treatment temperatures not lower than 620°C. Scanning electron microscopy (SEM) with energy-dispersive X-ray spectroscopy (EDX) analysis showed a compound uniformly composed of samarium, iron, arsenic, and oxygen (Sm-Fe-As-O) having surface areas on the order of 10 μm<sup>2</sup> surrounded by grains of SmAs, Fe<sub>2</sub>As, FeAs, and Sm<sub>2</sub>O<sub>3</sub> in a sample heated to 580°C. This work therefore shows the SmFeAsO phase grows at 620°C and suggested that the SmFeAsO phase emerge at 580°C.

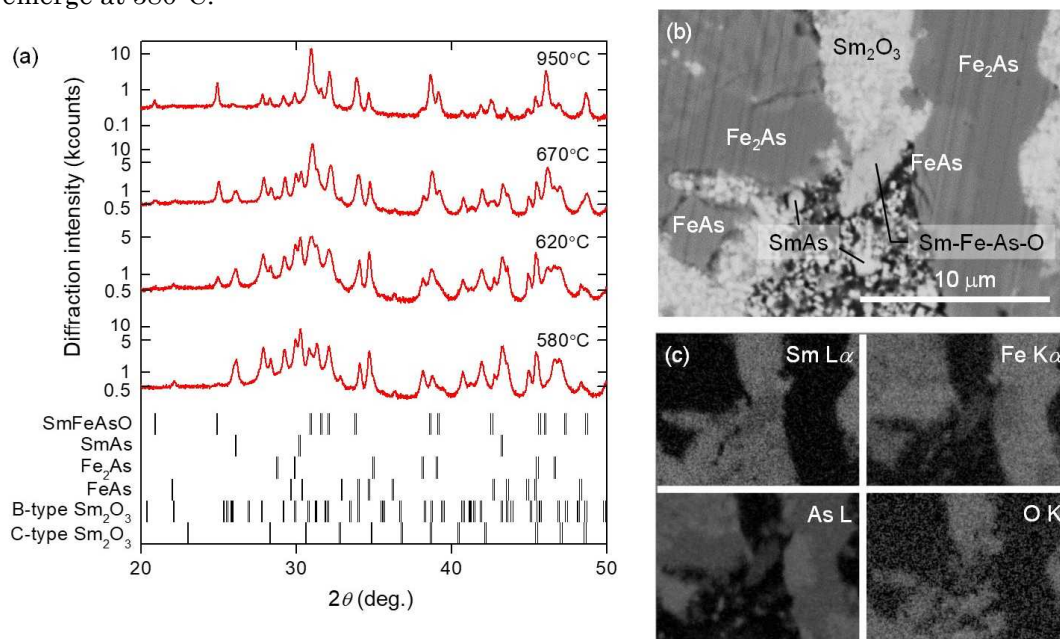


Fig. (a) Heat-treatment temperature dependence of X-ray diffraction (XRD) patterns at room temperature of the pulverized samples heated from SmAs, Fe<sub>2</sub>As, FeAs, and Sm<sub>2</sub>O<sub>3</sub> as starting materials. The vertical bars at the bottom represent diffractions due to SmFeAsO, SmAs, Fe<sub>2</sub>As, FeAs, B-type Sm<sub>2</sub>O<sub>3</sub>, and C-type Sm<sub>2</sub>O<sub>3</sub> from above. Heat-treatment temperatures are denoted near the patterns. (b) Back-scattered electron (BSE) scanning electron microscope (SEM) image and (c) energy-dispersive X-ray spectroscopy (EDX) elemental mapping of a polished sample heated to 580°C from SmAs, Fe<sub>2</sub>As, FeAs, and Sm<sub>2</sub>O<sub>3</sub> as starting materials.

[1] Y. Kamihara *et al.*, *J. Am. Chem. Soc.* **130**, 3296 (2008).

[2] Z.-A. Ren *et al.*, *Chin. Phys. Lett.* **25**, 2215 (2008).

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